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The sluggish ion diffusion and electrolyte freezing with volumetric changes limit the low-temperature performance of rechargeable batteries. Herein, a high-rate aqueous proton battery (APB) operated at and below -78 °C via a 62 wt% (9.5 m) H3PO4 electrolyte is reported. The APB is a rocking-chair battery that operates with protons commuting between a Prussian ...

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Phosphoric Acid Fuel Cell PAFC (Phosphoric Acid Fuel Cells) is a fuel cell with phosphoric acid as the electrolyte. In the anode, a hydrogen-rich gas containing CO 2 is passed, and the cathode is introduced with air as an oxidant. One of the distinguishing features of PAFCs is their resistance to CO 2, which makes PAFC the world"s first ground-based fuel cell [1, 2].

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The phosphoric acid diffusion coefficient through the acid doped membrane is about10-11m2s-1, at least one order of magnitude lower than that of aqueous phosphoric acid solutions.

The addition of phosphoric acid (PA) to the aqueous electrode slurry containing Li 4 Ti 5 O 12 (LTO) supresses the aluminum current collector corrosion and leads to a superior cycling performance of such electrodes in terms of de-/lithiation kinetics and cycling stability. These enable a very good rate capability of

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the utilized micrometer-sized LTO particles.

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Although phosphoric acid-doped polybenzimidazoles (PA-doped PBIs) are widely accepted in high-temperature proton exchange membrane fuel cells, further improvement is desirable to obtain optimal fuel cell performance. Block copolymers applied as low-temperature proton exchange membranes have been recently shown to exhibit high proton conductivity ...

Phosphoric acid (p-acid) is a key intermediate material in the production of lithium iron phosphate for the battery material supply chain. Currently there are two primary methods used in industry for the production of ...

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